

11-(3-Chloro-2-hydroxypropoxy)-2,3,9-trimethoxychromeno[3,4-*b*]chromen-12(6*H*)-one

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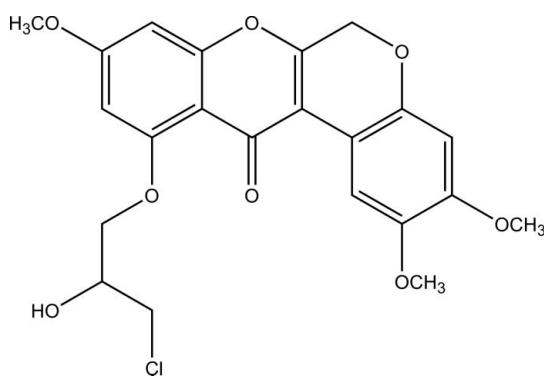
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Key indicators: single-crystal X-ray study; $T = 293\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$; disorder in main residue; R factor = 0.080; wR factor = 0.268; data-to-parameter ratio = 16.6.

In the title compound, $C_{22}H_{21}ClO_8$, the rotenoid core is nearly planar (r.m.s. deviation 0.114 Å), with the largest deviations from the least-squares plane being 0.286 (3) and 0.274 (2) Å. An intermolecular O—H···O hydrogen bond links two molecules into a centrosymmetric dimer having an $R_2^2(18)$ ring motif.

Related literature

For a related structure, see: Roengsumran *et al.* (2003).



Experimental

Crystal data

$C_{22}H_{21}ClO_8$	$\gamma = 74.455\text{ (3)}^\circ$
$M_r = 448.84$	$V = 1010.30\text{ (9)}\text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 7.1534\text{ (4)}\text{ \AA}$	Mo $K\alpha$ radiation
$b = 11.7904\text{ (6)}\text{ \AA}$	$\mu = 0.24\text{ mm}^{-1}$
$c = 12.7661\text{ (7)}\text{ \AA}$	$T = 293\text{ K}$
$\alpha = 76.901\text{ (3)}^\circ$	$0.30 \times 0.24 \times 0.20\text{ mm}$
$\beta = 86.991\text{ (3)}^\circ$	

Data collection

Bruker SMART APEXII
diffractometer
Absorption correction: none
14308 measured reflections

4517 independent reflections
2879 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.027$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.080$
 $wR(F^2) = 0.268$
 $S = 1.05$
4517 reflections
272 parameters

7 restraints
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.82\text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.91\text{ e \AA}^{-3}$

Table 1
Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O}8'\cdots\text{H}8'\cdots\text{O}8'^i$	0.82	2.2	2.81 (2)	132
$\text{O}8'\cdots\text{H}8'\cdots\text{O}7^i$	0.82	2.54	3.297 (10)	154
$\text{O}8\cdots\text{H}8\cdots\text{O}3^i$	0.82	1.92	2.735 (6)	169

Symmetry code: (i) $-x - 1, -y + 2, -z$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *publCIF* (Westrip, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2583).

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Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
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11-(3-Chloro-2-hydroxypropoxy)-2,3,9-trimethoxychromeno[3,4-*b*]chromen-12(6*H*)-one

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Comment

6-Deoxyclitoriacetal, extracted from the roots of *Stemona Collinsae* Craib, showed strong cytotoxic activity against various human carcinoma (Roengsumran *et al.*, 2003). In order to enhance its cytotoxic activities, the title compound was synthesized and its crystal structure was reported herein.

The bond lengths and angles in the molecules (Fig. 1) are within normal ranges and are comparable to a closely related structure (Roengsumran *et al.*, 2003). The rotenoid core is nearly flattened with the largest deviations from the least-squares plane of 0.286 (3) at C14 and -0.274 (2) at O2.

In the crystal structure, interatomic OH···O hydrogen bonds link the molecules into centrosymmetric dimers, forming $R_2^2(18)$ ring motifs, in which they may help in stabilize the crystal structure (Table 1).

Experimental

To the reaction mixture of 3-(4,5-dimethoxy-2-oxiranylmethoxy-phenyl)-3-hydroxy-7-methoxy-2-methyl-5-oxiranylmethoxy-chroman-4-one (20 mg, 0.046 mmol) in 3 ml of aqueous ethyl acetate was added 1*M* HCl (3 ml). The mixture solution was stirred for 30 min at room temperature. The reaction mixture was evaporated to remove solvent under reduced pressure. The residue was washed with water. Organic layer was extracted with dichloromethane and dried over MgSO₄. Solvent was removed under reduced pressure and the crude product was purified by silica gel column chromatography with dichloromethane:ethyl acetate:dichloromethane (2:1:2, *v/v*) as elutent to give the title compound as yellow crystal (70% yield).

Refinement

All non-H atoms were anisotropically refined. The O8 atom is disordered over two positions with site occupancies of 0.710 (5) and 0.290 (5). Restraints were used (SADI, DFIX and ISOR). The H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93 Å (aromatic), 0.97 Å (CH₂), 0.98 Å (CH₃) and O—H = 0.82 Å, and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}_{\text{aromatic}})$, $1.5U_{\text{eq}}(\text{C}_{\text{CH}_2})$, $1.5U_{\text{eq}}(\text{C}_{\text{CH}_3})$ and $1.2U_{\text{eq}}(\text{C}_\text{O})$, respectively.

Figures

Fig. 1. The molecular structure of the title compound, with disordered omitted. Displacement ellipsoids are drawn at 50% probability level.

Fig. 2. Packing diagram of A hydrogen bonded dimer of the title compound. Hydrogen bonds are shown as dashed lines.

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Crystal data

C ₂₂ H ₂₁ ClO ₈	V = 1010.30 (9) Å ³
M _r = 448.84	Z = 2
Triclinic, P $\bar{1}$	F ₀₀₀ = 468
Hall symbol: -P 1	D _x = 1.475 Mg m ⁻³
a = 7.1534 (4) Å	Mo K α radiation
b = 11.7904 (6) Å	λ = 0.71073 Å
c = 12.7661 (7) Å	θ = 1.6–27.4°
α = 76.901 (3)°	μ = 0.24 mm ⁻¹
β = 86.991 (3)°	T = 293 K
γ = 74.455 (3)°	Prism, colourless 0.30 × 0.24 × 0.20 mm

Data collection

Bruker SMART APEXII diffractometer	2879 reflections with $I > 2\sigma(I)$
Radiation source: Mo	R _{int} = 0.027
T = 295 K	θ _{max} = 27.4°
ω scans	θ _{min} = 1.6°
Absorption correction: none	h = -8→9
14308 measured reflections	k = -15→14
4517 independent reflections	l = -16→16

Refinement

Refinement on F^2	7 restraints
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)]$ = 0.080	w = 1/[σ ² (F _o) ² + (0.1402P) ² + 1.0019P] where P = (F _o ² + 2F _c ²)/3
wR(F ²) = 0.268	(Δ/σ) _{max} = 0.058
S = 1.05	Δρ _{max} = 0.82 e Å ⁻³
4517 reflections	Δρ _{min} = -0.91 e Å ⁻³
272 parameters	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
Cl1	-0.0091 (2)	0.94031 (14)	0.19387 (18)	0.1035 (7)	
O1	-0.8374 (4)	1.5195 (2)	0.1837 (2)	0.0440 (6)	
O2	-0.9242 (4)	1.7345 (2)	-0.0626 (2)	0.0530 (7)	
O3	-0.6183 (5)	1.2837 (2)	-0.0085 (2)	0.0565 (8)	
O4	-0.5594 (5)	1.5045 (3)	-0.3767 (2)	0.0589 (8)	
O5	-0.7253 (5)	1.7328 (3)	-0.4252 (2)	0.0598 (8)	
O6	-0.7703 (5)	1.2142 (3)	0.5002 (2)	0.0646 (9)	
O7	-0.5510 (5)	1.1103 (2)	0.1704 (2)	0.0566 (8)	
C21	-0.3641 (7)	0.9229 (3)	0.1446 (4)	0.063	
H21	-0.4543	0.9373	0.0849	0.075*	0.710 (5)
H21'	-0.3023	0.839	0.1856	0.075*	0.290 (5)
O8	-0.2980 (6)	0.7966 (4)	0.1808 (4)	0.064*	0.710 (5)
H8	-0.3167	0.7636	0.1337	0.096*	0.710 (5)
O8'	-0.5162 (14)	0.9032 (9)	0.0844 (8)	0.062*	0.290 (5)
H8'	-0.4975	0.9235	0.0198	0.093*	0.290 (5)
C1	-0.8078 (6)	1.3721 (4)	0.3395 (3)	0.0460 (9)	
H1	-0.8647	1.4325	0.376	0.055*	
C2	-0.7492 (6)	1.2532 (4)	0.3927 (3)	0.0475 (9)	
C3	-0.6659 (6)	1.1635 (4)	0.3367 (3)	0.0493 (9)	
H3	-0.6288	1.0833	0.3737	0.059*	
C4	-0.6377 (6)	1.1925 (3)	0.2272 (3)	0.0440 (9)	
C5	-0.6986 (5)	1.3151 (3)	0.1675 (3)	0.0368 (8)	
C6	-0.6830 (5)	1.3541 (3)	0.0510 (3)	0.0384 (8)	
C7	-0.7431 (5)	1.4857 (3)	0.0087 (3)	0.0334 (7)	
C8	-0.7322 (5)	1.5470 (3)	-0.1049 (3)	0.0346 (7)	
C9	-0.6415 (5)	1.4904 (3)	-0.1873 (3)	0.0383 (8)	
H9	-0.5796	1.4084	-0.1704	0.046*	
C10	-0.6422 (6)	1.5539 (3)	-0.2928 (3)	0.0418 (8)	
C11	-0.7324 (6)	1.6778 (3)	-0.3193 (3)	0.0427 (8)	
C12	-0.8198 (5)	1.7351 (3)	-0.2397 (3)	0.0437 (9)	
H12	-0.8786	1.8176	-0.2565	0.052*	
C13	-0.8204 (5)	1.6705 (3)	-0.1351 (3)	0.0392 (8)	
C14	-0.8613 (7)	1.6920 (3)	0.0448 (3)	0.0506 (10)	
H14A	-0.963	1.7257	0.091	0.061*	
H14B	-0.7485	1.7195	0.0543	0.061*	
C15	-0.8106 (5)	1.5572 (3)	0.0779 (3)	0.0386 (8)	
C16	-0.7789 (5)	1.3987 (3)	0.2294 (3)	0.0389 (8)	
C17	-0.4440 (7)	1.3837 (4)	-0.3532 (3)	0.0577 (11)	
H17A	-0.3951	1.3606	-0.4189	0.087*	
H17B	-0.5212	1.3315	-0.3172	0.087*	
H17C	-0.3373	1.3769	-0.3075	0.087*	
C18	-0.8071 (8)	1.8595 (4)	-0.4555 (4)	0.0658 (13)	
H18A	-0.7916	1.8867	-0.5315	0.099*	
H18B	-0.7423	1.8994	-0.4172	0.099*	
H18C	-0.9427	1.878	-0.4384	0.099*	

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C19	-0.8712 (9)	1.3025 (5)	0.5581 (4)	0.0774 (15)
H19A	-0.8778	1.2646	0.6326	0.116*
H19B	-1.0002	1.3384	0.5293	0.116*
H19C	-0.8034	1.3639	0.5512	0.116*
C20	-0.4775 (7)	0.9901 (3)	0.2280 (4)	0.0574 (11)
H20A	-0.5823	0.9553	0.2579	0.069*
H20B	-0.3926	0.9867	0.2861	0.069*
C22	-0.1968 (9)	0.9713 (4)	0.0981 (5)	0.0854 (19)
H22A	-0.2427	1.0579	0.0714	0.102*
H22B	-0.1453	0.9357	0.0377	0.102*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0831 (11)	0.0732 (10)	0.1646 (18)	-0.0261 (8)	0.0238 (10)	-0.0457 (11)
O1	0.0532 (15)	0.0392 (14)	0.0383 (14)	-0.0067 (11)	0.0102 (11)	-0.0147 (11)
O2	0.0622 (18)	0.0374 (14)	0.0484 (16)	0.0066 (12)	0.0055 (13)	-0.0121 (12)
O3	0.087 (2)	0.0332 (14)	0.0452 (16)	-0.0064 (13)	0.0152 (14)	-0.0156 (12)
O4	0.093 (2)	0.0470 (16)	0.0332 (14)	-0.0100 (15)	0.0080 (13)	-0.0128 (12)
O5	0.084 (2)	0.0487 (16)	0.0385 (15)	-0.0118 (15)	-0.0030 (14)	-0.0003 (12)
O6	0.078 (2)	0.072 (2)	0.0392 (16)	-0.0197 (16)	0.0067 (14)	-0.0040 (15)
O7	0.078 (2)	0.0320 (14)	0.0488 (16)	-0.0001 (13)	0.0084 (14)	-0.0057 (12)
C21	0.085	0.033	0.063	0	-0.011	-0.011
C1	0.050 (2)	0.053 (2)	0.040 (2)	-0.0170 (17)	0.0080 (16)	-0.0177 (17)
C2	0.048 (2)	0.057 (2)	0.040 (2)	-0.0202 (18)	0.0058 (16)	-0.0085 (18)
C3	0.048 (2)	0.045 (2)	0.051 (2)	-0.0122 (17)	0.0055 (17)	-0.0033 (18)
C4	0.046 (2)	0.0394 (19)	0.044 (2)	-0.0087 (15)	0.0068 (15)	-0.0075 (16)
C5	0.0376 (18)	0.0336 (17)	0.0400 (19)	-0.0086 (13)	0.0051 (14)	-0.0117 (14)
C6	0.0443 (19)	0.0310 (17)	0.0410 (19)	-0.0091 (14)	0.0063 (15)	-0.0124 (15)
C7	0.0331 (17)	0.0309 (16)	0.0372 (17)	-0.0072 (12)	0.0040 (13)	-0.0121 (14)
C8	0.0335 (17)	0.0331 (17)	0.0386 (18)	-0.0091 (13)	0.0018 (13)	-0.0107 (14)
C9	0.047 (2)	0.0303 (16)	0.0366 (18)	-0.0076 (14)	0.0031 (14)	-0.0095 (14)
C10	0.049 (2)	0.0412 (19)	0.0370 (19)	-0.0118 (15)	0.0010 (15)	-0.0116 (15)
C11	0.049 (2)	0.0403 (19)	0.0364 (19)	-0.0126 (16)	-0.0054 (15)	-0.0024 (15)
C12	0.046 (2)	0.0349 (18)	0.046 (2)	-0.0046 (15)	-0.0035 (16)	-0.0057 (16)
C13	0.0401 (19)	0.0341 (18)	0.0427 (19)	-0.0053 (14)	0.0017 (14)	-0.0126 (15)
C14	0.062 (3)	0.0362 (19)	0.049 (2)	-0.0019 (17)	0.0034 (18)	-0.0157 (17)
C15	0.0379 (18)	0.0354 (18)	0.0417 (19)	-0.0064 (14)	0.0038 (14)	-0.0114 (15)
C16	0.0381 (18)	0.0401 (19)	0.0404 (19)	-0.0105 (14)	0.0043 (14)	-0.0135 (15)
C17	0.079 (3)	0.051 (2)	0.044 (2)	-0.013 (2)	0.015 (2)	-0.0207 (19)
C18	0.086 (3)	0.050 (2)	0.052 (3)	-0.013 (2)	-0.007 (2)	0.006 (2)
C19	0.102 (4)	0.093 (4)	0.038 (2)	-0.025 (3)	0.011 (2)	-0.018 (2)
C20	0.065 (3)	0.039 (2)	0.061 (3)	-0.0100 (18)	0.003 (2)	-0.0015 (19)
C22	0.117 (5)	0.037 (2)	0.079 (4)	0.004 (2)	0.039 (3)	-0.007 (2)

Geometric parameters (\AA , $^\circ$)

Cl1—C22	1.772 (7)	C5—C16	1.389 (5)
O1—C15	1.343 (4)	C5—C6	1.462 (5)

O1—C16	1.369 (4)	C6—C7	1.473 (5)
O2—C13	1.386 (4)	C7—C15	1.342 (5)
O2—C14	1.402 (5)	C7—C8	1.474 (5)
O3—C6	1.233 (4)	C8—C13	1.394 (5)
O4—C10	1.367 (5)	C8—C9	1.410 (5)
O4—C17	1.415 (5)	C9—C10	1.383 (5)
O5—C11	1.366 (4)	C9—H9	0.93
O5—C18	1.420 (5)	C10—C11	1.400 (5)
O6—C2	1.359 (5)	C11—C12	1.379 (6)
O6—C19	1.428 (6)	C12—C13	1.379 (5)
O7—C4	1.342 (5)	C12—H12	0.93
O7—C20	1.412 (5)	C14—C15	1.496 (5)
C21—O8	1.410 (5)	C14—H14A	0.97
C21—O8'	1.460 (7)	C14—H14B	0.97
C21—C22	1.500 (8)	C17—H17A	0.96
C21—C20	1.538 (6)	C17—H17B	0.96
C21—H21	0.98	C17—H17C	0.96
C21—H21'	1.0012	C18—H18A	0.96
O8—H21'	0.5101	C18—H18B	0.96
O8—H8	0.82	C18—H18C	0.96
O8'—H8'	0.82	C19—H19A	0.96
C1—C2	1.373 (6)	C19—H19B	0.96
C1—C16	1.386 (5)	C19—H19C	0.96
C1—H1	0.93	C20—H20A	0.97
C2—C3	1.394 (6)	C20—H20B	0.97
C3—C4	1.379 (6)	C22—H22A	0.97
C3—H3	0.93	C22—H22B	0.97
C4—C5	1.435 (5)		
C15—O1—C16	119.0 (3)	O4—C10—C11	115.4 (3)
C13—O2—C14	115.8 (3)	C9—C10—C11	119.8 (3)
C10—O4—C17	118.1 (3)	O5—C11—C12	124.9 (3)
C11—O5—C18	117.9 (3)	O5—C11—C10	115.7 (3)
C2—O6—C19	117.0 (4)	C12—C11—C10	119.4 (3)
C4—O7—C20	117.3 (3)	C13—C12—C11	120.2 (3)
O8—C21—O8'	88.0 (5)	C13—C12—H12	119.9
O8—C21—C22	109.0 (4)	C11—C12—H12	119.9
O8'—C21—C22	126.5 (6)	C12—C13—O2	115.7 (3)
O8—C21—C20	115.0 (4)	C12—C13—C8	122.5 (3)
O8'—C21—C20	103.5 (5)	O2—C13—C8	121.7 (3)
C22—C21—C20	112.9 (4)	O2—C14—C15	112.2 (3)
O8—C21—H21	106.4	O2—C14—H14A	109.2
O8'—C21—H21	22.8	C15—C14—H14A	109.2
C22—C21—H21	106.4	O2—C14—H14B	109.2
C20—C21—H21	106.4	C15—C14—H14B	109.2
O8—C21—H21'	14.8	H14A—C14—H14B	107.9
O8'—C21—H21'	101.7	O1—C15—C7	125.7 (3)
C22—C21—H21'	104.5	O1—C15—C14	111.6 (3)
C20—C21—H21'	105.8	C7—C15—C14	122.7 (3)
H21—C21—H21'	121	O1—C16—C5	121.0 (3)

supplementary materials

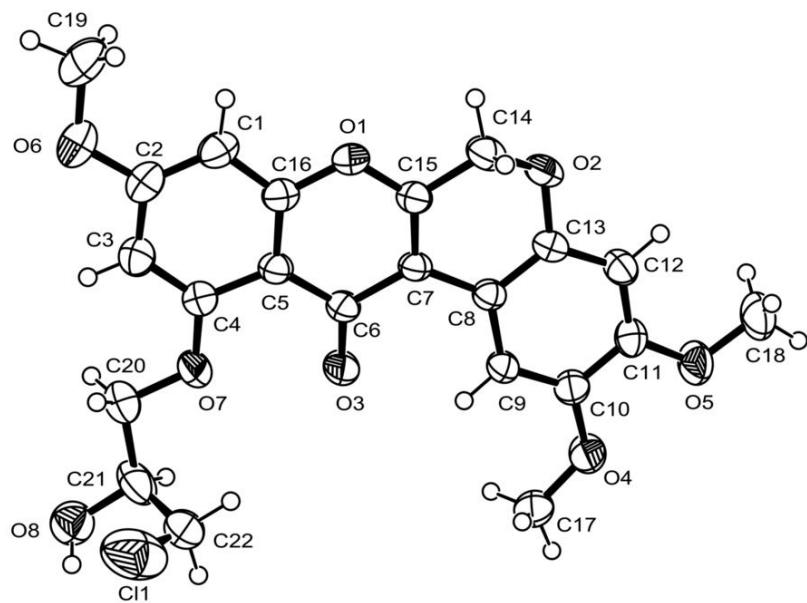
C21—O8—H21'	30	O1—C16—C1	113.4 (3)
C21—O8—H8	109.5	C5—C16—C1	125.6 (3)
H21'—O8—H8	138.7	O4—C17—H17A	109.5
C21—O8'—H8'	109.5	O4—C17—H17B	109.5
C2—C1—C16	117.5 (4)	H17A—C17—H17B	109.5
C2—C1—H1	121.2	O4—C17—H17C	109.5
C16—C1—H1	121.2	H17A—C17—H17C	109.5
O6—C2—C1	123.8 (4)	H17B—C17—H17C	109.5
O6—C2—C3	115.7 (4)	O5—C18—H18A	109.5
C1—C2—C3	120.6 (4)	O5—C18—H18B	109.5
C4—C3—C2	120.8 (4)	H18A—C18—H18B	109.5
C4—C3—H3	119.6	O5—C18—H18C	109.5
C2—C3—H3	119.6	H18A—C18—H18C	109.5
O7—C4—C3	123.1 (3)	H18B—C18—H18C	109.5
O7—C4—C5	116.0 (3)	O6—C19—H19A	109.5
C3—C4—C5	120.9 (3)	O6—C19—H19B	109.5
C16—C5—C4	114.6 (3)	H19A—C19—H19B	109.5
C16—C5—C6	120.4 (3)	O6—C19—H19C	109.5
C4—C5—C6	125.0 (3)	H19A—C19—H19C	109.5
O3—C6—C5	123.4 (3)	H19B—C19—H19C	109.5
O3—C6—C7	121.5 (3)	O7—C20—C21	104.9 (3)
C5—C6—C7	115.1 (3)	O7—C20—H20A	110.8
C15—C7—C8	116.3 (3)	C21—C20—H20A	110.8
C15—C7—C6	118.4 (3)	O7—C20—H20B	110.8
C8—C7—C6	125.2 (3)	C21—C20—H20B	110.8
C13—C8—C9	116.4 (3)	H20A—C20—H20B	108.8
C13—C8—C7	118.4 (3)	C21—C22—C11	112.2 (4)
C9—C8—C7	125.2 (3)	C21—C22—H22A	109.2
C10—C9—C8	121.7 (3)	C11—C22—H22A	109.2
C10—C9—H9	119.1	C21—C22—H22B	109.2
C8—C9—H9	119.1	C11—C22—H22B	109.2
O4—C10—C9	124.7 (3)	H22A—C22—H22B	107.9

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O8'—H8'···O8 ⁱ	0.82	2.2	2.81 (2)	132
O8'—H8'···O7 ⁱ	0.82	2.54	3.297 (10)	154
O8—H8···O3 ⁱ	0.82	1.92	2.735 (6)	169

Symmetry codes: (i) $-x-1, -y+2, -z$.

Fig. 1



supplementary materials

Fig. 2

